

38. Larutan garam X bertindak balas dengan larutan zink nitrat,  $\text{Zn}(\text{NO}_3)_2$  menghasilkan pepejal berwarna putih. Antara berikut, manakah yang betul bagi menguji kehadiran anion bagi garam X?

*Salt solution X reacts with zinc nitrate,  $\text{Zn}(\text{NO}_3)_2$  solution to produce a white solid.  
Which of the following is correct to test the presence of anion for salt X?*

- A Tambah  $2 \text{ cm}^3$  HCl diikuti dengan  $2 \text{ cm}^3$   $\text{BaCl}_2$  ke dalam larutan X  
*Add  $2 \text{ cm}^3$  HCl followed by  $2 \text{ cm}^3$   $\text{BaCl}_2$  into X solution*
- B Tambah  $2 \text{ cm}^3$  HCl ke dalam larutan X dan alirkan gas terbebas dlm air kapur  
*Add  $2 \text{ cm}^3$  HCl into X solution and flow the liberated gas into lime water*
- C Tambah  $2 \text{ cm}^3$   $\text{H}_2\text{SO}_4$  cair,  $2 \text{ cm}^3$  larutan  $\text{FeSO}_4$  dan beberapa titis  $\text{H}_2\text{SO}_4$  pekat ke dalam larutan X  
*Add  $2 \text{ cm}^3$  dilute  $\text{H}_2\text{SO}_4$ ,  $2 \text{ cm}^3$   $\text{FeSO}_4$  solution and a few drops concentrated  $\text{H}_2\text{SO}_4$  into X solution*
- D Tambah  $2 \text{ cm}^3$   $\text{H}_2\text{SO}_4$  dan  $2 \text{ cm}^3$   $\text{MgSO}_4$  ke dalam larutan X  
*Add  $2 \text{ cm}^3$   $\text{H}_2\text{SO}_4$  and  $2 \text{ cm}^3$   $\text{MgSO}_4$  into X solution*